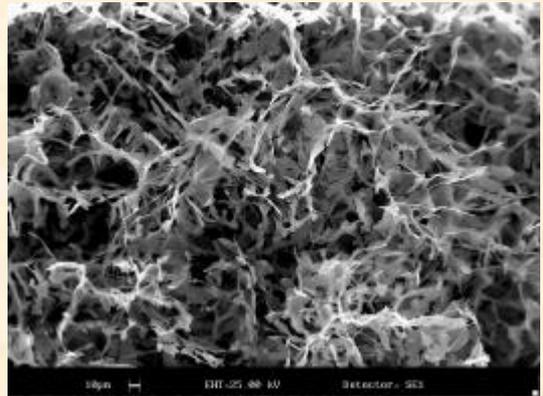


¹Netherlands Institute for Metals Research (NIMR), Delft, The Netherlands.

²Laboratory for Material Science, Delft University of Technology, Delft, The Netherlands.

1. Introduction

The figure on the right shows the isolated intermetallic skeleton of *b*-Al₅FeSi, from an as cast 6005A series aluminium alloy, where the Al is dissolved by the Sibut method. The *b*-particles have a thin plate like shape, which adversely influence the extrusion properties of the aluminum. Therefore the material is homogenised, where the *b*- intermetallics transform to the more favorable *a*-Al₁₂(Fe_xMn_{1-x})₃Si particles. To understand the nature of this transformation, the changes have been investigated by optical microscopy, SEM and Energy Dispersive Spectrography (EDS).



b-Al₅FeSi plates in as cast material

2. The kinetics of the phase transformation

As cast

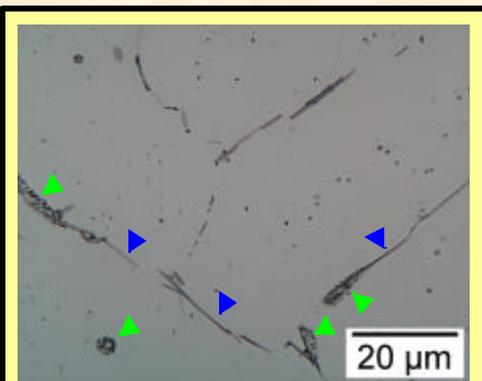


Plate like *b*-Al₅FeSi particles (▼) inside and on the grain boundaries.
(Mg₂Si precipitates (▼) are nucleated near the boundary of the *b*-particles and in clusters.)

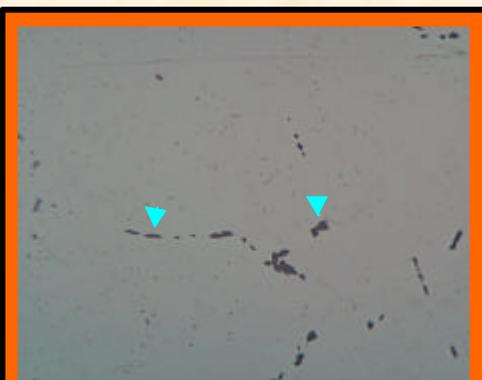
(Optical microscopy pictures on polished samples)

Intermediate transformation: 2 hours homogenised (540 °C)



Nucleation of *a*-Al₁₂(Fe_xMn_{1-x})₃Si particles (▼) on the *b*-Al₅FeSi particles.
(Mg₂Si precipitates are dissolved in the Al matrix)

32 hours homogenised (540 °C)



b-Al₅FeSi particles are completely transformed to the more rounded *a*-Al₁₂(Fe_xMn_{1-x})₃Si particles (▼)

Alloy composition of 6005 series aluminium in wt%

Mg	Si	Fe	Mn	Ti,Cu,Zn,other
0.69	0.88	0.29	0.18	<0.03

3. Quantitative measurements

The transformation is measured quantitatively by automatic SEM and Energy Dispersive Spectrography (EDS) measurements. To accomplish this, 300 intermetallics taken at random are analysed on each polished sample.

3.1 EDS measurements

Figure 1 shows the EDS measurements on particles in cast material, where the particles with Mg are ignored. The measured aluminium concentration varies because in general a part of the Al-matrix was also measured by EDS. However, this figure shows that the iron to silicon ratio of the intermetallics was approximately unity, which corresponds to the expected $b\text{-Al}_5\text{FeSi}$ stoichiometry.

$b\text{-Al}_5\text{FeSi}$ is the main phase of intermetallics in the cast structure.

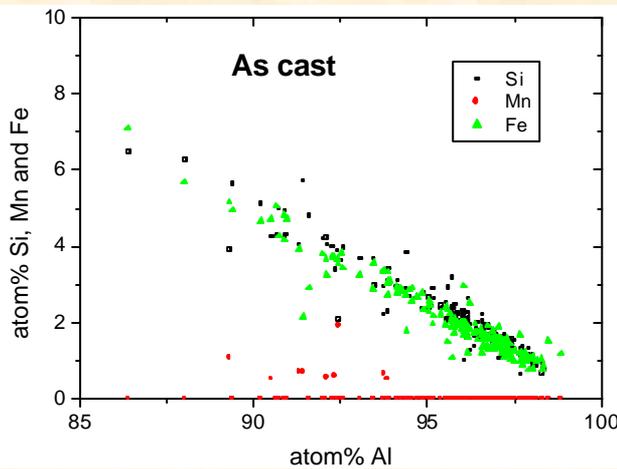


Figure 1, assembly of EDS measurements.

The data collected from EDS measurements (see Figure 2) on samples which have been homogenised for longer times (32 hours at 590 °C) shows that the [Fe]:[Mn]:[Si] ratio is ~2:1:1

$a\text{-Al}_{12}(\text{Fe}_x\text{Mn}_{1-x})_3\text{Si}$ was the main phase of intermetallics after homogenisation.

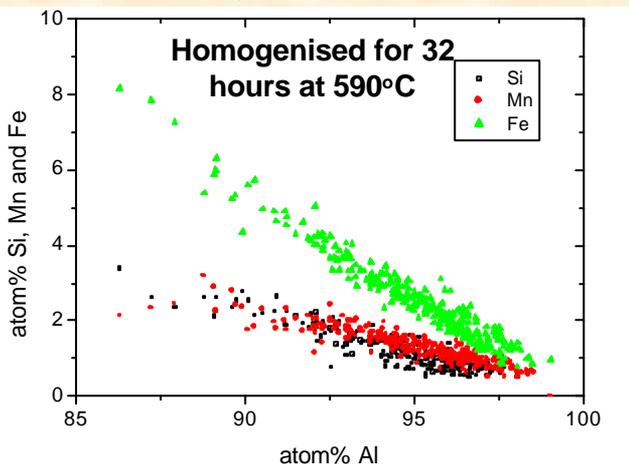


Figure 2, assembly of EDS measurements.

3.2 Transformation rate

Figure 3 shows the relative volume% as function of time of $a\text{-Al}_5\text{FeSi}$ and $b\text{-Al}_{12}(\text{Fe}_x\text{Mn}_{1-x})_3\text{Si}$ particles, as measured by the automatic SEM and EDS.

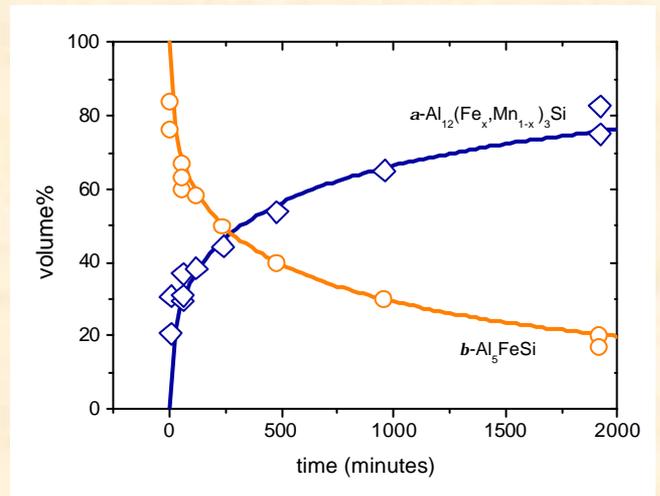


Figure 3. The relative volume% of *a* and *b* as a function of time.

The line fitted through the experimental data for volume% of *a* in Figure 3 is a plot of the Avrami equation:

$$x_a = 1 - \exp(-kt^n)$$

Where k is a constant and n in this case is $n \approx 0.5$, which indicates that the transformation is controlled by diffusion over the observed time range. Other measurements indicated that the transformation rate was significantly accelerated when the temperature was higher (565 °C and 590 °C).

4. Discussion and conclusion

Optical microscopy

Nuclei of $a\text{-Al}_{12}(\text{Fe}_x\text{Mn}_{1-x})_3\text{Si}$ particles are found on $b\text{-Al}_5\text{FeSi}$ particles

at the very beginning of the transformation, the transformation rate is nucleation controlled.

SEM and EDS measurements

Avrami plots

the transformation kinetics for higher times (>60 minutes) is diffusion controlled.